

# DOLOMITE FUM Australia(Franchi Umberto Marmi)

Chemwatch Hazard Alert Code: 0

Chemwatch: 21551

Issue Date: 04/03/2024 Version No: 8.1 Print Date: 04/03/2024 Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements S.GHS.AUS.EN.E

### SECTION 1 Identification of the substance / mixture and of the company / undertaking

#### **Product Identifier**

Product name	DOLOMITE	
Chemical Name	Not Available	
Synonyms	aCO3, MgCO3; magnesium calcium carbonate; calcium magnesium carbonate; calcium magnesium dicarbonate (CaMg(CO3)2); CAS RN: 9598-19-2); carbonic acid, calcium magnesium salt (2:1:1); carbonic acid, calcium magnesium salt, basic,; Dolomite Lime; Maglime	
Chemical formula	C-H2-O3 .1/2 Ca .1/2 Mg	
Other means of identification	Not Available	
CAS number	16389-88-1	

#### Relevant identified uses of the substance or mixture and uses advised against

Refractory for furnaces, manufacture of magnesium compounds and magnesium metal. As building material, in fertilizers, stock feeds, paper Relevant identified uses making, ceramics, mineral wool, removal of sulfur dioxide from stack gases. As a welding flux ingredient.

#### Details of the manufacturer or supplier of the safety data sheet

Registered company name	FUM Australia(Franchi Umberto Marmi)	
Address	99-101, Fairbank Road Clayton South VIC 3169 Australia	
Telephone	03 9546 3644	
Fax	Not Available	
Website	Not Available	
Email	Not Available	

#### Emergency telephone number

Association / Organisation	Not Available
Emergency telephone numbers	Not Available
Other emergency telephone numbers	Not Available

#### **SECTION 2 Hazards identification**

#### Classification of the substance or mixture

Poisons Schedule	Not Applicable
Classification <sup>[1]</sup>	Not Applicable

#### Label elements

Hazard pictogram(s)	Not Applicable
Signal word	Not Applicable

#### Hazard statement(s)

Not Applicable

Precautionary statement(s) Prevention Not Applicable Precautionary statement(s) Response Not Applicable Precautionary statement(s) Storage Not Applicable Precautionary statement(s) Disposal Not Applicable

### DOLOMITE

#### Substances

CAS No	%[weight]	Name	
16389-88-1	>97	dolomite	
Not Available		mixture calcium and magnesium (Ca CO. Mg CO3)	
7631-86-9	0-3	silica amorphous	
7732-18-5	<2	water	
	by Chemwatch; 2. Classification	drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C	

#### **Mixtures**

See section above for composition of Substances

### **SECTION 4 First aid measures**

### Description of first aid measures

Eye Contact	<ul> <li>If this product comes in contact with eyes:</li> <li>Wash out immediately with water.</li> <li>If irritation continues, seek medical attention.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>	
Skin Contact	If skin or hair contact occurs: <ul> <li>Flush skin and hair with running water (and soap if available).</li> <li>Seek medical attention in event of irritation.</li> </ul>	
Inhalation	Inhalation       If dust is inhaled, remove from contaminated area.         Encourage patient to blow nose to ensure clear passage of breathing.         If irritation or discomfort persists seek medical attention.	
Ingestion	<ul> <li>Immediately give a glass of water.</li> <li>First aid is not generally required. If in doubt, contact a Poisons Information Centre or a doctor.</li> </ul>	

### Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

### **SECTION 5 Firefighting measures**

### Extinguishing media

- There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

### Special hazards arising from the substrate or mixture

Fire Incompatibility None known.

Advice for firefighters Fire Fighting	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear breathing apparatus plus protective gloves in the event of a fire.</li> <li>Prevent, by any means available, spillage from entering drains or water courses.</li> <li>Use fire fighting procedures suitable for surrounding area.</li> <li>DO NOT approach containers suspected to be hot.</li> <li>Cool fire exposed containers with water spray from a protected location.</li> <li>If safe to do so, remove containers from path of fire.</li> <li>Equipment should be thoroughly decontaminated after use.</li> </ul>
Fire/Explosion Hazard	<ul> <li>Non combustible.</li> <li>Not considered a significant fire risk, however containers may burn.</li> <li>Decomposition may produce toxic fumes of: metal oxides</li> <li>May emit poisonous fumes.</li> </ul>
HAZCHEM	Not Applicable

### **SECTION 6 Accidental release measures**

### Personal precautions, protective equipment and emergency procedures

See section 8

#### **Environmental precautions**

See section 12

### Methods and material for containment and cleaning up

- Clean up all spills immediately.
  - Avoid breathing dust and contact with skin and eyes.
- Wear protective clothing, gloves, safety glasses and dust respirator.
- Minor Spills
  - Sweep up, shovel up or
    - ▶ Vacuum up (consider explosion-proof machines designed to be grounded during storage and use).
    - Place spilled material in clean, dry, sealable, labelled container.

### DOLOMITE

Major Spills
--------------

Personal Protective Equipment advice is contained in Section 8 of the SDS.

# SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	<ul> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of exposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>Prevent concentration in hollows and sumps.</li> <li>DO NOT enter confined spaces until atmosphere has been checked.</li> <li>DO NOT allow material to contact humans, exposed food or food utensils.</li> <li>Avoid contact with incompatible materials.</li> <li>When handling, DO NOT eat, drink or smoke.</li> <li>Keep containers securely sealed when not in use.</li> <li>Avoid physical damage to containers.</li> <li>Always wash hands with soap and water after handling.</li> <li>Work clothes should be laundered separately. Launder contaminated clothing before re-use.</li> <li>Use good occupational work practice.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.</li> </ul>
Other information	<ul> <li>Store in original containers.</li> <li>Keep containers securely sealed.</li> <li>Store in a cool, dry area protected from environmental extremes.</li> <li>Store away from incompatible materials and foodstuff containers.</li> <li>Protect containers against physical damage and check regularly for leaks.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>For major quantities:</li> <li>Consider storage in bunded areas - ensure storage areas are isolated from sources of community water (including stormwater, ground water, lakes and streams).</li> <li>Ensure that accidental discharge to air or water is the subject of a contingency disaster management plan; this may require consultation with local authorities.</li> </ul>

# Conditions for safe storage, including any incompatibilities

Suitable container       Polyethylene or polypropylene container.         Check all containers are clearly labelled and free from leaks.	
Storage incompatibility	<ul> <li>Metals and their oxides or salts may react violently with chlorine trifluoride and bromine trifluoride.</li> <li>These trifluorides are hypergolic oxidisers. They ignite on contact (without external source of heat or ignition) with recognised fuels - contact with these materials, following an ambient or slightly elevated temperature, is often violent and may produce ignition.</li> <li>The state of subdivision may affect the results.</li> </ul>

# SECTION 8 Exposure controls / personal protection

### **Control parameters**

### Occupational Exposure Limits (OEL)

INGREDIENT	ΔΤΔ

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	silica amorphous	Silica - Amorphous: Precipitated silica	10 mg/m3	Not Available	Not Available	(a) This value is for inhalable dust containing no asbestos and < 1% crystalline silica.
Australia Exposure Standards	silica amorphous	Silica - Amorphous: Silica gel	10 mg/m3	Not Available	Not Available	<ul> <li>(a) This value is for inhalable dust containing no asbestos and &lt; 1% crystalline silica.</li> </ul>
Australia Exposure Standards	silica amorphous	Silica - Amorphous: Diatomaceous earth (uncalcined)	10 mg/m3	Not Available	Not Available	(a) This value is for inhalable dust containing no asbestos and < 1% crystalline silica.
Australia Exposure Standards	silica amorphous	Silica - Amorphous: Fumed silica (respirable dust)	2 mg/m3	Not Available	Not Available	Not Available
Australia Exposure Standards	silica amorphous	Silica, fused	0.05 mg/m3	Not Available	Not Available	Not Available
Australia Exposure Standards	silica amorphous	Silica - Amorphous: Fume (thermally generated)(respirable dust)	2 mg/m3	Not Available	Not Available	(e) Containing no asbestos and < 1% crystalline silica.

# DOLOMITE

Ingredient	TEEL-1	TEEL-2			TEEL-3
silica amorphous	18 mg/m3	200 mg/m3			1,200 mg/m3
silica amorphous	18 mg/m3	18 mg/m3 100 mg/m3			630 mg/m3
silica amorphous	120 mg/m3	1,300 mg/m3			7,900 mg/m3
silica amorphous	45 mg/m3	500 mg/m3			3,000 mg/m3
silica amorphous	18 mg/m3	740 mg/m3	740 mg/m3		4,500 mg/m3
Ingredient	Original IDLH		Rev	ised IDLH	
dolomite	Not Available	Not Available		Not Available	
silica amorphous	3,000 mg/m3	3,000 mg/m3		Not Available	
water	Not Available		Not	Available	

# Exposure controls

<ul> <li>Engineering controls are used to remove a hazard or place a be highly effective in protecting workers and will typically be if The basic types of engineering controls are:</li> <li>Process controls which involve changing the way a job activit Enclosure and/or isolation of emission source which keeps a "adds" and "removes" air in the work environment. Ventilation system must match the particular process and che Employers may need to use multiple types of controls to prev</li> <li>Local exhaust ventilation is required where solids are han proportion will be powdered by mutual friction.</li> <li>If in spite of local exhaust an adverse concentration of the Such protection might consist of:</li> <li>(a): particle dust respirators, if necessary, combined with an a (b): filter respirators with absorption cartridge or canister of the (c): fresh-air hoods or masks.</li> <li>Air contaminants generated in the workplace possess varying circulating air required to effectively remove the contaminant.</li> </ul>	independent of worker interactions to provide this high level ty or process is done to reduce the risk. selected hazard "physically" away from the worker and von can remove or dilute an air contaminant if designed pro- emical or contaminant in use. went employee overexposure. Indled as powders or crystals; even when particulates are the substance in air could occur, respiratory protection sho absorption cartridge; he right type; g "escape" velocities which, in turn, determine the "capture	vel of protection. ventilation that strategically perly. The design of a relatively large, a certain uld be considered.
of very high rapid air motion). Within each range the appropriate value depends on: Lower end of the range 1: Room air currents minimal or favourable to capture 2: Contaminants of low toxicity or of nuisance value only. 3: Intermittent, low production. 4: Large hood or large air mass in motion Simple theory shows that air velocity falls rapidly with distance with the square of distance from the extraction point (in simpl accordingly, after reference to distance from the contaminatin 4-10 m/s (800-2000 f/min) for extraction of crusher dusts gen	conveyer loading, crusher dusts, gas discharge (active nerated dusts (released at high initial velocity into zone Upper end of the range 1: Disturbing room air currents 2: Contaminants of high toxicity 3: High production, heavy use 4: Small hood-local control only are away from the opening of a simple extraction pipe. Vel le cases). Therefore the air speed at the extraction point is ng source. The air velocity at the extraction fan, for example reated 2 metres distant from the extraction point. Other m	should be adjusted, ple, should be a minimum of nechanical considerations,
the wearing of lenses or restrictions on use, should be cr and adsorption for the class of chemicals in use and an a their removal and suitable equipment should be readily a remove contact lens as soon as practicable. Lens should	reated for each workplace or task. This should include a r account of injury experience. Medical and first-aid person available. In the event of chemical exposure, begin eye irr d be removed at the first signs of eye redness or irritation	review of lens absorption nel should be trained in rigation immediately and - lens should be removed in
See Hand protection below		
<ul> <li>manufacturer. Where the chemical is a preparation of several and has therefore to be checked prior to the application.</li> <li>The exact break through time for substances has to be obtain making a final choice.</li> <li>Personal hygiene is a key element of effective hand care. Glo washed and dried thoroughly. Application of a non-perfumed</li> </ul>	Il substances, the resistance of the glove material can not ned from the manufacturer of the protective gloves and he oves must only be worn on clean hands. After using glove moisturiser is recommended.	t be calculated in advance as to be observed when
	<ul> <li>grinding, abrasive blasting, tumbling, high speed wheel ge of very high rapid air motion).</li> <li>Within each range the appropriate value depends on: <ul> <li>Lower end of the range</li> <li>1: Room air currents minimal or favourable to capture</li> <li>2: Contaminants of low toxicity or of nuisance value only.</li> <li>3: Intermittent, low production.</li> <li>4: Large hood or large air mass in motion</li> </ul> </li> <li>Simple theory shows that air velocity falls rapidly with distance with the square of distance from the extraction point (in simp accordingly, after reference to distance from the contamination 4-10 m/s (800-2000 f/min) for extraction of crusher dusts ger producing performance deficits within the extraction apparate more when extraction systems are installed or used.</li> <li>Image: Safety glasses with side shields</li> <li>Chemical goggles.</li> <li>Contact lenses may pose a special hazard; soft contact the wearing of lenses or restrictions on use, should be caud adsorption for the class of chemicals in use and an their removal and suitable equipment should be readily a remove contact lens as soon as practicable. Lens should a clean environment only after workers have washed ha national equivalent]"</li> <li>See Hand protection below</li> <li>The selection of suitable gloves does not only depend on the manufacturer. Where the chemical is a preparation of severa and has therefore to be checked prior to the application. The exact break through time for substances has to be obtair making a final choice.</li> <li>Personal hygiene is a key element of effective hand care. Gl washed and dried thoroughly. Application of a non-perfumed Suitability and durability of glove type is dependent on usage of requency and duration of contact,</li> <li>chemical resistance of glove material,</li> </ul>	grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).         Within each range the appropriate value depends on:       Lower end of the range       Upper end of the range         1: Room air currents minimal or favourable to capture       1: Disturbing room air currents         2: Contaminants of low toxicity or of nuisance value only.       2: Contaminants of high toxicity         3: Intermittent, low production.       3: High production, heavy use         4: Large hood or large air mass in motion       4: Small hood-local control only         Simple theory shows that ir velocity falls rapidly with distance away from the opening of a simple extraction point. Other r         accordingly, after reference to distance from the contaminating source. The air velocity at the extraction point. Other r         producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are mul more when extraction systems are installed or used.         View       A: Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written pot the wearing of lenses or restrictions on use, should be readily available. In the vertaci exposure, begin yeig in remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin national equivalent]*         See Hand protection below       The selection of suitable gloves does not only depend on the m

<ul> <li>Thinner gloves (down to 0.1 mm or less) may be required where a high degree of manual dexterity is needed. However, these gloves are only likely to give short duration protection and would normally be just for single use applications, then disposed of.</li> <li>Thicker gloves (up to 3 mm or more) may be required where there is a mechanical (as well as a chemical) risk i.e. where there is abrasion or puncture potential Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.</li> <li>Experience indicates that the following polymers are suitable as glove materials for protection against undissolved, dry solids, where abrasive particles are not present.</li> <li>polychloroprene.</li> <li>nitrile rubber.</li> <li>butyl rubber.</li> <li>fluorocaoutchouc.</li> <li>polyvinyl chloride.</li> <li>Gloves should be examined for wear and/ or degradation constantly.</li> <li>Body protection</li> <li>See Other protection below</li> </ul>	Body protection	<ul> <li>likely to give short duration protection and would normally be just for single use applications, then disposed of.</li> <li>Thicker gloves (up to 3 mm or more) may be required where there is a mechanical (as well as a chemical) risk i.e. where there is abrasion or puncture potential</li> <li>Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.</li> <li>Experience indicates that the following polymers are suitable as glove materials for protection against undissolved, dry solids, where abrasive particles are not present.</li> <li>polychloroprene.</li> <li>initrile rubber.</li> <li>butyl rubber.</li> <li>fluorocaoutchouc.</li> <li>polyvinyl chloride.</li> <li>Gloves should be examined for wear and/ or degradation constantly.</li> </ul>
Other protection <ul> <li>P.V.C apron.</li> <li>Barrier cream.</li> <li>Skin cleansing cream.</li> <li>Eye wash unit.</li> <li>Eye wash unit.</li> </ul>	Other protection	<ul> <li>P.V.C apron.</li> <li>Barrier cream.</li> <li>Skin cleansing cream.</li> </ul>

### Recommended material(s)

GLOVE SELECTION INDEX

### Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection:

DOLOMITE

Material	CPI
BUTYL	А
NEOPRENE	А
VITON	А
NATURAL RUBBER	С
PVA	С

\* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

**NOTE:** As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

#### Ansell Glove Selection

Glove — In order of recommendation	
AlphaTec 02-100	
AlphaTec® Solvex® 37-185	
AlphaTec® 38-612	
AlphaTec® 58-008	
AlphaTec® 58-530B	
NphaTec® 58-530W	
AlphaTec® 58-735	
AlphaTec® 79-700	
NphaTec® Solvex® 37-675	

**Respiratory protection** 

Type -P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	P1 Air-line*	-	PAPR-P1 -
up to 50 x ES	Air-line**	P2	PAPR-P2
up to 100 x ES	-	P3	-
		Air-line*	-
100+ x ES	-	Air-line**	PAPR-P3

\* - Negative pressure demand \*\* - Continuous flow

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.

 The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure - ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option).

 Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.

 Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory protection program.

 Where protection from nuisance levels of dusts are desired, use type N95 (US) or type P1 (EN143) dust masks. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU)

Use approved positive flow mask if significant quantities of dust becomes airborne.
 Try to avoid creating dust conditions.

Continued...

Page 6 of 10

### DOLOMITE

### DermaShield™ 73-711

The suggested gloves for use should be confirmed with the glove supplier.

# **SECTION 9 Physical and chemical properties**

### Information on basic physical and chemical properties

		· · · / · II · II · · · · · · · · · · ·	
Appearance	Gray - white solid. No odour. Slight solubility in wa magnesium and calcium carbonates is CAS RN: 1		e mineral, dolomite, magnesium as the principle ore, and
Physical state	Divided Solid	Relative density (Water = 1)	2.9 approx.
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Applicable
pH (as supplied)	Not Applicable	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Applicable
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Applicable	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Applicable
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Partly miscible	pH as a solution (1%)	9.6 (50% solution)
Vapour density (Air = 1)	Not Applicable	VOC g/L	Not Applicable

# **SECTION 10 Stability and reactivity**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

### **SECTION 11 Toxicological information**

### Information on toxicological effects

Inhaled	The material is not thought to produce adverse health effects or irritation of the respiratory tract (as classified by EC Directives using animal models). Nevertheless, good hygiene practice requires that exposure be kept to a minimum and that suitable control measures be used in an occupational setting. Persons with impaired respiratory function, airway diseases and conditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled. If prior damage to the circulatory or nervous systems has occurred or if kidney damage has been sustained, proper screenings should be conducted on individuals who may be exposed to further risk if handling and use of the material result in excessive exposures.
Ingestion	The material has <b>NOT</b> been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroborating animal or human evidence. Not normally a hazard due to physical form of product. Considered an unlikely route of entry in commercial/industrial environments
Skin Contact	The material is not thought to produce adverse health effects or skin irritation following contact (as classified by EC Directives using animal models). Nevertheless, good hygiene practice requires that exposure be kept to a minimum and that suitable gloves be used in an occupational setting. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.
Eye	Although the material is not thought to be an irritant (as classified by EC Directives), direct contact with the eye may cause transient discomfort characterised by tearing or conjunctival redness (as with windburn). Slight abrasive damage may also result.
Chronic	Dolomite is generally inert and non-toxic. However, as with any mined product particulates can carry bacteria into the passage and lungs, producing infection and bronchitis. No specific toxic effects. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure. Long term exposure to high dust concentrations may cause changes in lung function i.e. pneumoconiosis, caused by particles less than 0.5 micron penetrating and remaining in the lung. There has been some concern that this material can cause cancer or mutations but there is not enough data to make an assessment.

deless fr	ΤΟΧΙΟΙΤΥ	IRRITATION	
dolomite	Not Available	Not Available	
	ΤΟΧΙΟΙΤΥ	IRRITATION	
	dermal (rat) LD50: >2000 mg/kg <sup>[1]</sup>	Eye (rabbit): no	n-irritating ** [Grace]
silica amorphous	Inhalation(Rat) LC50: >0.09<0.84 mg/l4h <sup>[1]</sup>	Eye: no adverse	e effect observed (not irritating) <sup>[1]</sup>
	Oral (Rat) LD50: >1000 mg/kg <sup>[1]</sup>	Skin (rabbit): no	n-irritating *
		Skin: no advers	e effect observed (not irritating) <sup>[1]</sup>
	ΤΟΧΙΟΙΤΥ	IRRITATION	
water	Oral (Rat) LD50: >90000 mg/kg <sup>[2]</sup>	Not Available	
Legend:	1. Value obtained from Europe ECHA Registered Su specified data extracted from RTECS - Register of To		ined from manufacturer's SDS. Unless otherwise
SILICA AMORPHOUS	effects were reversible. [PATTYS] For silica amorphous: Derived No Adverse Effects Level (NOAEL) in the rat In humans, synthetic amorphous silica (SAS) is esse evidence of adverse health effects due to SAS. Repe drying/cracking of the skin. When experimental animals inhale synthetic amorph vast majority of SAS is excreted in the faeces and thu- via urine without modification in animals and humans After ingestion, there is limited accumulation of SAS but appears to be insignificant in animals and humans indication of metabolism of SAS in animals or human soluble in physiological media and the soluble chemi Both the mammalian and environmental toxicology o of solubility and particle size. SAS has no acute intrir were caused by the presence of high numbers of res representative of exposure to commercial SASs and cause dryness and cracking, SAS is not a skin or eye Repeated-dose and chronic toxicity studies confirm t Long-term inhalation of SAS caused some adverse e which subsided after exposure. Numerous repeated-dose, subchronic and chronic in concentrations ranging from 0.5 mg/m3 to 150 mg/m mg/m3. When available, the no-observed adverse effect explained by different particle size, and therefore the does the NOAEL/LOAEL. Neither inhalation nor oral administration caused nec assays. SAS does not impair development of the foe were not affected. For Synthetic Amorphous Silica (SAS) Repeated dose toxicity Oral (rat), 2 weeks to 6 months, no significant treatm Inhalation (rat), 13 weeks, Lowest Observed Effects in For silane treated synthetic amorphous silica: Repeated dose toxicity: oral (rat), 28-d, diet, no signi There is no evidence of cancer or other long-term res SAS. Respiratory symptoms in SAS workers have be function values and chest radiographs are not adverse The substance is classified by IARC as Group 3: <b>NOT</b> classifiable as to its carcinogenicity to humans. Evidence of carcinogenicity may be inadequate or lin No significant acute toxicological data identified in lift	ntially non-toxic by mouth, skin or eye- rated exposure (without personal prote- bus silica (SAS) dust, it dissolves in the rere is little accumulation in the body. F SAS is not expected to be broken do in body tissues and rapid elimination of s. SASs injected subcutaneously are si- s based on chemical structure and av- cal species that are formed are elimina f SASs are significantly influenced by the should not be used for human risk asso- irritant, and it is not a sensitiser. The absence of toxicity when SAS is sw ffects in animals (increases in lung infl- halation toxicity studies have been cor 3. Lowest-observed adverse effect leve feet levels (NOAELs) were between 0. number of particles administered per plasms (tumours). SAS is not mutage tus. Fertility was not specifically studies the lungs and effects in the nasal cav ficant treatment-related adverse effects at doses or evel (LOEL) =1.3 mg/m3 based on mil to the lungs and effects (for example, s een shown to correlate with smoking br- seen shown to correlate with smoking br- seely affected by long-term exposure to hited in animal testing.	cition) may cause mechanical irritation of the eye and e lung fluid and is rapidly eliminated. If swallowed, th ollowing absorption across the gut, SAS is eliminated wn (metabolised) in mammals. ccurs. Intestinal absorption has not been calculated, subjected to rapid dissolution and removal. There is to allable data. In contrast to crystalline silica, SAS is ated via the urinary tract without modification. the physical and chemical properties, particularly tho cts, including suffocation, that have been reported e required test atmosphere. These results are not sessment. Though repeated exposure of the skin ma rallowed or upon skin contact. lammation, cell injury and lung collagen content), all nducted with SAS in a number of species, at airborne els (LOAELs) were typically in the range of 1 to 50 5 and 10 mg/m3. The difference in values may be unit dose. In general, as particle size decreases so nic in vitro. No genotoxicity was detected in in vivo d, but the reproductive organs in long-term studies f up to 8% silica in the diet. d reversible effects in the lungs. Inhalation (rat), 90 ity. s at the doses tested. lifcosis) in workers employed in the manufacture of ut not with SAS exposure, while serial pulmonary
Acute Toxicity	×	Carcinogenicity	×
Skin Irritation/Corrosion	×	Reproductivity	×
Serious Eye Damage/Irritation	×	STOT - Single Exposure	×
Respiratory or Skin	×	STOT - Repeated Exposure	×
sensitisation			<b>^</b>

Data available to make classification

# **SECTION 12 Ecological information**

Toxicity					
	Endpoint	Test Duration (hr)	Species	Value	Source
dolomite	Not Available	Not Available	Not Available	Not Available	Not Available

silica amorphous	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50	48h	Crustacea	>86mg/l	2
	EC50	96h	Algae or other aquatic plants	217.576mg/l	2
	EC50	72h	Algae or other aquatic plants	14.1mg/l	2
	EC0(ECx)	24h	Crustacea	>=10000mg/l	1
	LC50	96h	Fish	1033.016mg/l	2
	Endpoint	Test Duration (hr)	Species	Value	Source
water	Not Available	Not Available	Not Available	Not Available	Not Available
Legend:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data				

#### For Metal:

Atmospheric Fate - Metal-containing inorganic substances generally have negligible vapour pressure and are not expected to partition to air.

Environmental Fate: Environmental processes, such as oxidation, the presence of acids or bases and microbiological processes, may transform insoluble metals to more soluble ionic forms. Environmental processes may enhance bioavailability and may also be important in changing solubilities.

Aquatic/Terrestrial Fate: When released to dry soil, most metals will exhibit limited mobility and remain in the upper layer; some will leach locally into ground water and/ or surface water ecosystems when soaked by rain or melt ice. A metal ion is considered infinitely persistent because it cannot degrade further. Once released to surface waters and moist soils their fate depends on solubility and dissociation in water. A significant proportion of dissolved/ sorbed metals will end up in sediments through the settling of suspended particles. The remaining metal ions can then be taken up by aquatic organisms. Ionic species may bind to dissolved ligands or sorb to solid particles in water.

Ecotoxicity: Even though many metals show few toxic effects at physiological pH levels, transformation may introduce new or magnified effects.

**DO NOT** discharge into sewer or waterways.

#### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
silica amorphous	LOW	LOW
water	LOW	LOW

#### **Bioaccumulative potential**

Ingredient	Bioaccumulation	
silica amorphous	LOW (LogKOW = 0.5294)	

### Mobility in soil

Ingredient	Mobility
silica amorphous	LOW (KOC = 23.74)

#### **SECTION 13 Disposal considerations**

Waste treatment methods	
Product / Packaging disposal	Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked. A Hierarchy of Controls seems to be common - the user should investigate: • Reduction • Reduction • Recycling • Disposal (if all else fails) This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate. In most instances the supplier of the material should be consulted. • DO NOT allow wash water from cleaning or process equipment to enter drains. • It may be necessary to collect all wash water for treatment before disposal. • In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first. • Where in doubt contact the responsible authority. • Recycle wherever possible or consult manufacturer for recycling options. • Consult State Land Waste Management Authority for disposal. • Bury residue in an authorised landfill. • Recycle containers if possible, or dispose of in an authorised landfill.

### **SECTION 14 Transport information**

Labels Required	
Marine Pollutant	NO
HAZCHEM	Not Applicable

### Land transport (ADG): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

# 14.7.1. Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

### 14.7.2. Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
dolomite	Not Available
silica amorphous	Not Available
water	Not Available

# 14.7.3. Transport in bulk in accordance with the IGC Code

Product name	Ship Type
dolomite	Not Available
silica amorphous	Not Available
water	Not Available

# **SECTION 15 Regulatory information**

### Safety, health and environmental regulations / legislation specific for the substance or mixture

### dolomite is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

### silica amorphous is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals

Australia Model Work Health and Safety Regulations - Hazardous chemicals (other than lead) requiring health monitoring

Australian Inventory of Industrial Chemicals (AIIC)

Chemical Footprint Project - Chemicals of High Concern List

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs - Not Classified as Carcinogenic

International WHO List of Proposed Occupational Exposure Limit (OEL) Values for Manufactured Nanomaterials (MNMS)

# water is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

### Additional Regulatory Information

Not Applicable

### **National Inventory Status**

National Inventory	Status	
Australia - AIIC / Australia Non-Industrial Use	Yes	
Canada - DSL	Yes	
Canada - NDSL	No (water)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	No (dolomite)	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	Yes	
Taiwan - TCSI	Yes	
Mexico - INSQ	Yes	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

# **SECTION 16 Other information**

Revision Date	04/03/2024
Initial Date	03/01/2002

## SDS Version Summary

Version	Date of Update	Sections Updated
7.1	03/09/2017	Identification of the substance / mixture and of the company / undertaking - Synonyms
8.1	20/06/2022	Expiration. Review and Update

Chemwatch: 21551	Page <b>10</b> of <b>10</b>	Issue Date: 04/03/2024
Version No: 8.1	DOLOMITE	Print Date: 04/03/2024

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### **Definitions and abbreviations**

- PC TWA: Permissible Concentration-Time Weighted Average
- PC STEL: Permissible Concentration-Short Term Exposure Limit
- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists
- STEL: Short Term Exposure Limit
- TEEL: Temporary Emergency Exposure Limit.
- ► IDLH: Immediately Dangerous to Life or Health Concentrations
- ES: Exposure Standard
- OSF: Odour Safety Factor
- NOAEL: No Observed Adverse Effect Level
   LOAEL: Lowest Observed Adverse Effect Level
- TLV: Threshold Limit Value
- LOD: Limit Of Detection
- OTV: Odour Threshold Value
- BCF: BioConcentration Factors
- BEI: Biological Exposure Index
- DNEL: Derived No-Effect Level
- PNEC: Predicted no-effect concentration
- AIIC: Australian Inventory of Industrial Chemicals
- Alic. Australian Inventory of Industrial Chemil
   DSL: Domestic Substances List
- NDSL: Non-Domestic Substances List
- IECSC: Inventory of Existing Chemical Substance in China
- EINECS: European INventory of Existing Commercial chemical Substances
- ELINCS: European List of Notified Chemical Substances
- NLP: No-Longer Polymers
- ENCS: Existing and New Chemical Substances Inventory
- KECI: Korea Existing Chemicals Inventory
- NZIoC: New Zealand Inventory of Chemicals
- PICCS: Philippine Inventory of Chemicals and Chemical Substances
- TSCA: Toxic Substances Control Act
- TCSI: Taiwan Chemical Substance Inventory
- INSQ: Inventario Nacional de Sustancias Químicas
- NCI: National Chemical Inventory
- FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

### This document is copyright.

Apart from any fair dealing for the purposes of private study, research, review or criticism, as permitted under the Copyright Act, no part may be reproduced by any process without written permission from CHEMWATCH.

TEL (+61 3) 9572 4700.